

2-(2,4-Dihydroxybenzoyl)benzoic acid

Yang Li,^{a*} Feng-Yan Ge,^a
Li-Gong Chen,^a Chuan-Ming
Dong,^a Xi-Long Yan,^a Er-Hong
Duan,^b Tao Zeng,^a Yue-Cheng
Zhang^a and Guo-Yi Bai^c

^aCollege of Pharmaceuticals and Biotechnology, Tianjin University, Tianjin 300072, People's Republic of China, ^bResearch and Development Center for Petrochemical Technology, Tianjin University, Tianjin 300072, People's Republic of China, and ^cCollege of Chemistry and Environmental Science, Hebei University, Baoding 071002, People's Republic of China

Correspondence e-mail: liyang777@tju.edu.cn

Key indicators

Single-crystal X-ray study

$T = 294$ K

Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å

R factor = 0.049

wR factor = 0.146

Data-to-parameter ratio = 14.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title structure, $\text{C}_{14}\text{H}_{10}\text{O}_5$, the angle between the planes formed by the 2,4-dihydroxybenzoyl and *o*-benzoic acid moieties is $87.12(4)^\circ$. In addition to an intramolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond, intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds ($\text{H}\cdots\text{O} = 1.76$ and 1.89 Å) connect molecules to form a two-dimensional network parallel to $(10\bar{1})$.

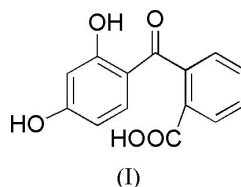
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Comment

The title compound, (I), is an intermediate in the synthesis of fluorescein and was first prepared by Baeyer (1876) and also by Bollmann (1922) in his study of resorcinbenzein. Further details about the title compound and its derivatives have been reported (Orndorff & Adamson, 1918; Orndorff & Kelley, 1922; Orndorff & Kline, 1924). Despite extensive investigations with respect to its synthesis, there has not been a crystallographic study of (I). The present study reports the crystal structure of 2-(2,4-dihydroxybenzoyl)benzoic acid at room temperature.



Selected bond lengths and angles for (I) are given in Table 1. The 2,4-dihydroxybenzoyl and *o*-benzoic acid moieties are each essentially planar, with maximum deviations from each plane of $0.0170(21)$ Å for C5 and $0.0417(17)$ Å for O5, and the angle between these planes is $86.79(4)^\circ$. In addition to an intramolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond, intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds connect molecules to form a two-dimensional network parallel to $(10\bar{1})$ (see Table 2 and Fig. 2).

Experimental

The title compound was prepared according to the method described by Orndorff & Kline (1924). Crystals suitable for X-ray diffraction were obtained by slow evaporation of a solution in methanol and water. ^1H NMR ($\text{DMSO}-d_6$): δ 6.21 (*dd*, $^4J = 2.0$ Hz, $^3J = 8.8$ Hz, 1H), 6.32 (*d*, $J = 2.4$ Hz, 1H), 6.92 (*d*, $J = 8.8$ Hz, 1H), 7.36 (*dd*, $^4J = 0.8$ Hz, $^3J = 7.6$ Hz, 1H), 7.58–7.70 (*m*, 2H), 8.08 (*dd*, $^4J = 0.8$ Hz, $^3J = 8.0$ Hz, 1H), 10.68 (*s*, 1H), 12.22 (*s*, 1H), 13.15 (*s*, 1H); ^{13}C NMR ($\text{DMSO}-d_6$): δ 103.63, 109.03, 114.95, 128.63, 130.66, 131.48, 133.42, 136.06, 142.00, 166.59, 168.70, 202.82.

Crystal data

C₁₄H₁₀O₅
M_r = 258.22
 Monoclinic, *P*2₁/*n*
a = 10.331 (3) Å
b = 11.628 (4) Å
c = 11.640 (4) Å
 β = 116.034 (5)°
V = 1256.4 (7) Å³
Z = 4

D_x = 1.365 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 2633 reflections
 θ = 2.6–26.5°
 μ = 0.11 mm⁻¹
T = 294 (2) K
 Block, colourless
 0.30 × 0.22 × 0.22 mm

Data collection

Bruker SMART CCD diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Bruker, 1997)
T_{min} = 0.963, *T_{max}* = 0.977
 6755 measured reflections

2626 independent reflections
 1762 reflections with *I* > 2σ(*I*)
R_{int} = 0.099
 θ_{max} = 26.7°
h = -12 → 12
k = -13 → 14
l = -14 → 13

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.049
wR (*F*²) = 0.146
S = 1.07
 2626 reflections
 184 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0478P)^2 + 0.3516P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.27 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$

Table 1 Selected geometric parameters (Å, °).

| | | | |
|-------------|-------------|--------------|-------------|
| O2—C1 | 1.316 (2) | C3—C8 | 1.513 (3) |
| O4—C10 | 1.353 (3) | C8—C9 | 1.436 (3) |
| C1—C2 | 1.488 (3) | | |
| O1—C1—O2 | 122.6 (2) | O3—C8—C9 | 121.81 (17) |
| C4—C3—C8 | 116.49 (17) | C14—C9—C8 | 121.67 (17) |
| C2—C3—C8 | 124.45 (17) | C10—C9—C8 | 121.12 (18) |
| O1—C1—C2—C7 | -179.5 (2) | O2—C1—C2—C3 | 178.9 (2) |
| O2—C1—C2—C7 | 0.0 (3) | O3—C8—C9—C14 | 178.77 (19) |
| O1—C1—C2—C3 | -0.7 (3) | O3—C8—C9—C10 | -0.5 (3) |

Table 2 Hydrogen-bonding geometry (Å, °).

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|--------------------------|-------------|---------------|-----------------------|-------------------------|
| O2—H2...O3 ⁱ | 0.89 (3) | 1.76 (3) | 2.643 (2) | 177 (3) |
| O4—H4...O3 | 0.90 (3) | 1.80 (3) | 2.602 (2) | 148 (3) |
| O5—H5...O1 ⁱⁱ | 0.89 (4) | 1.89 (4) | 2.764 (2) | 169 (3) |

Symmetry codes: (i) $\frac{1}{2} + x, \frac{1}{2} - y, \frac{1}{2} + z$; (ii) $-x, 1 - y, 1 - z$.

All H atoms bonded to C atoms were included in calculated positions, with C—H = 0.93 Å. They were included in the refinement in riding-model approximation, with *U*_{iso}(H) = 1.2*U*_{eq}(C). The H atoms bonded to O atoms were refined independently with isotropic displacement parameters.

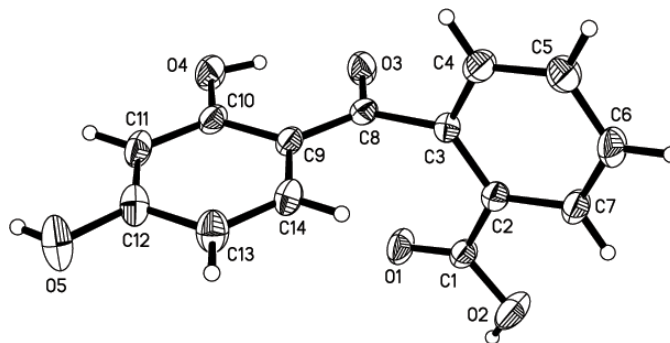


Figure 1 A view of the molecular of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

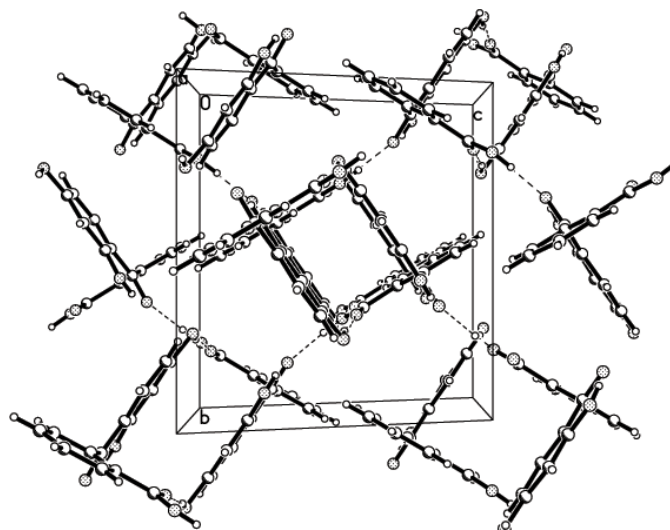


Figure 2 The molecular structure of (I), viewed along the *a* axis. Dashed lines indicate hydrogen-bond interactions.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

References

Baeyer (1876). *Justus Liebigs Ann. Chem.* **183**, 23–24.
 Bollmann, F. (1922). *J. Prakt. Chem.* **104**, 123–126.
 Bruker (1997). *SADABS, SMART, SAINT and SHELXTL* (Version 5.10). Bruker AXS Inc., Madison, Wisconsin, USA.
 Orndorff, W. R. & Adamson, W. A. (1918). *J. Am. Chem. Soc.* **40**, 1235–1257.
 Orndorff, W. R. & Kelley, L. (1922). *J. Am. Chem. Soc.* **44**, 1518–1527.
 Orndorff, W. R. & Kline, E. (1924). *J. Am. Chem. Soc.* **46**, 2276–2291.
 Sheldrick, G. M. (1997). *SHELXS97 and SHELXL97*. University of Göttingen, Germany.